

NATIONAL EXAMINATIONS DECEMBER 2019

16-Mec-A1 Applied Thermodynamics and Heat Transfer

3 Hours Duration

Notes :

1. If doubt exists concerning the interpretation of any question, the candidate is urged to make assumptions and clearly explain what has been assumed along with the answer to the question.
2. The examination is open book. As a consequence, candidates are permitted to make use of any textbooks, references or notes.
3. Any non-communicating calculator is permitted. However, candidates must indicate the type of calculator(s) that they have used by writing the name and model designation of the calculator(s) on the inside of the cover of the first examination book.
4. It is expected that each candidate will have copies of both a thermodynamics text and a heat transfer text in order to make use of the information presented in the tables and graphs contained.
5. The answers to five questions, either three questions from Part A and two questions from Part B or two questions from Part A and three questions from Part B, comprise a complete examination.
6. Candidates must indicate the answers that they wish to have graded on the cover of the first examination book. Otherwise the answers will be graded in the order in which they appear in the examination book(s) up to a maximum of three answers per section .
7. The answer to any question carries the same value in the grading .

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PART A - THERMODYNAMICS

1(a) A cylinder filled with 50 gm of air at 1 atmosphere pressure and a temperature of 20°C is adiabatically but irreversibly compressed to 10 atmospheres pressure. When equilibrium has been established, the final temperature is found to be 320°C . Determine how much work was done on the system and the change in entropy during the compression. If a reversible process had occurred between the same two state points, would heat have been added or removed? Compute the value of the exponent in the relationship $pV^n = \text{Constant}$ describing the irreversible compression process? Plot the reversible and irreversible processes on p-v and T-s diagrams.

(b) A refrigeration unit whose volume is 0.050 m^3 is evacuated and then slowly charged with Freon 12. The effect of different charging processes are to be considered. The temperature of the refrigeration unit may be assumed to be 20°C for each of the questions posed below.

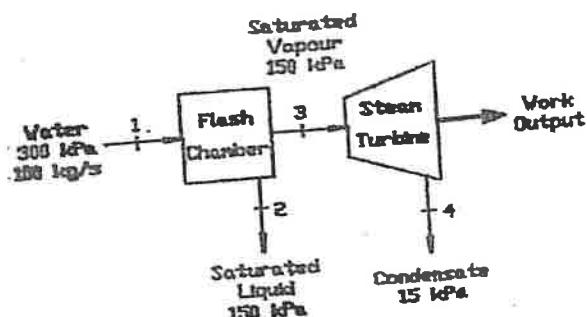
What would the mass of the Freon 12 in the refrigeration unit be if the system pressure were 250 kPa?

What would the mass of the Freon 12 in the refrigeration unit be if it were filled with saturated Freon 12?

What would the mass fractions of vapour and liquid in the refrigeration unit be if it contained 5.0 kg of Freon 12?

How much heat would have to be removed if the 5.0 kg of Freon 12 in the refrigeration unit entered at 0.60 MPa and 30°C ?

2. Saturated water at 300 kPa enters a flash chamber at the rate of 100 kg/s. Saturated liquid at 150 kPa exits from the bottom of the flash chamber and saturated vapour at 150 kPa exits from the top of the flash chamber. The vapour stream flows to a steam turbine having an isentropic efficiency of 90% from which it exits at a pressure of 15 kPa. For steady state operation, negligible heat transfer with the surroundings and insignificant kinetic and potential energy effects, determine the power produced by the turbine. Sketch the processes on a Ts diagram.



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3. An air compressor having a 75 mm bore and 100 mm stroke operates at 800 revolutions per minute. The compressor receives air at atmospheric pressure and a temperature of 27°C and delivers it at a pressure of 412 kPa. The volumetric efficiency of the compressor is 80%. Assuming the thermal efficiency of the compression process to be 90%, determine the power required to drive the compressor.

4. A vapour-compression system having Freon 12 as the working fluid is to be used as a heat pump to extract 23.5 kW from the ambient air outside a house at a temperature of 0°C in order to transfer it to the air circulating inside of the house at 25°C. There is a difference of 10°C between the evaporator temperature and the outside air temperature as well as a difference of 7.5°C between the condenser temperature and the inside air temperature. For the purposes of the analysis, assume that the vapour entering the compressor is saturated, that the liquid at the inlet of the expansion valve is saturated. Determine the coefficient of performance and the saving to be realized from the use of the heat pump if electricity is available at 4.5 cents per kilowatt hour.

PART B - HEAT TRANSFER

5. An evacuated thick walled cylindrical tube 2 cm inside diameter by 5 cm outside diameter surrounds a concentrically located high temperature radiant heater which produces a 10^3 W/m^2 heat flux at the inner surface. The temperature of the outer surface of the tube, which is cooled by a flowing fluid at 300 K, is 633 K. The thermal conductivity of the tube material is $2.2 \text{ W/m}^\circ\text{C}$. Determine

- (a) The rate of heat transfer per unit length of cylindrical tube.
- (b) The heat transfer coefficient at the outer surface of the tube.
- (c) The temperature of the inner surface of the cylindrical tube.

Prove that the temperature of the inner surface of the cylindrical tube would decrease if 1 cm of material with thermal conductivity $\kappa = 5 \text{ W/m}^\circ\text{C}$ were applied at the outer surface of the tube.

6. Heat is generated in a 75 mm diameter cylindrical electrical conductor made of a material having a thermal conductivity of $70 \text{ W/m}^\circ\text{C}$ which is located in a fluid having a temperature of 27°C. The surface heat transfer coefficient is $568 \text{ W/m}^2\text{ }^\circ\text{C}$ and the temperature must not exceed 540°C anywhere in the cylinder.

- (a) Determine the maximum heat generation rate per unit length.
- (b) Determine the surface temperature under these conditions.

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7. During aluminum arc welding, molten metal droplets are ejected. While some of these droplets might be hot enough to ignite and burn, most droplets simply cool down. Suppose that a particular molten droplet has a diameter $D = 0.5 \text{ mm}$, an initial temperature $T_i = 1700 \text{ K}$ and an initial velocity $V = 1 \text{ m/s}$. Determine the rate at which it cools in air at $T_\infty = 300 \text{ K}$ due to the loss of heat by convection and radiation. The properties of the molten droplet are estimated to be $\rho = 2100 \text{ kg/m}^3$, $C_p = 1100 \text{ J/kgK}$ and $\epsilon = 0.20$.

8. Hot oil ($C_p = 2200 \text{ J/kg}^\circ\text{C}$) is to be cooled by water ($C_p = 4180 \text{ J/kg}^\circ\text{C}$) in a 2 shell-pass 12 tube-pass heat exchanger. The tubes are thin walled and made of copper with internal diameter of 1.8 cm. The length of each tube pass in the heat exchanger is 3 m and the overall heat transfer coefficient is 340 $\text{W/m}^2\text{ }^\circ\text{C}$. The water flows through the tubes at 0.1 kg/s and the oil flows through the shell at 0.2 kg/s. The water and the oil enter at 18°C and 160°C respectively. Determine the rate of heat transfer and the temperatures at which the water and the oil exit the heat exchanger.

The End

Thermodynamic Properties of Freon-12 (Dichlorodifluoromethane)*

Saturated Freon-12

Temp. °C	Abs. Press. MPa <i>P</i>	Specific Volume m ³ /kg			Enthalpy kJ/kg			Entropy kJ/kg K		
		Sat. Liquid <i>v_f</i>	Evap. <i>v_g</i>	Sat. Vapor <i>v_g</i>	Sat. Liquid <i>h_f</i>	Evap. <i>h_g</i>	Sat. Vapor <i>h_g</i>	Sat. Liquid <i>s_f</i>	Evap. <i>s_g</i>	Sat. Vapor <i>s_g</i>
-90	0.0028	0.000 608	4.414 937	4.415 545	-43.243	189.618	146.375	-0.2084	1.0352	0.8268
-85	0.0042	0.000 612	3.036 704	3.037 316	-38.968	187.608	148.640	-0.1854	0.9970	0.8116
-80	0.0062	0.000 617	2.137 728	2.138 345	-34.688	185.612	150.924	-0.1630	0.9609	0.7979
-75	0.0088	0.000 622	1.537 030	1.537 651	-30.401	183.625	153.224	-0.1411	0.9266	0.7855
-70	0.0123	0.000 627	1.126 654	1.127 280	-26.103	181.640	155.536	-0.1197	0.8940	0.7744
-65	0.0168	0.000 632	0.840 534	0.841 166	-21.793	179.651	157.857	-0.0987	0.8630	0.7643
-60	0.0226	0.000 637	0.637 274	0.637 910	-17.469	175.641	160.184	-0.0782	0.8334	0.7552
-55	0.0300	0.000 642	0.490 358	0.491 000	-13.129	173.611	162.512	-0.0581	0.8051	0.7470
-50	0.0391	0.000 648	0.382 457	0.383 105	-8.772	171.558	164.840	-0.0384	0.7779	0.7396
-45	0.0504	0.000 654	0.302 029	0.302 682	-4.396	169.479	167.163	-0.0190	0.7519	0.7329
-40	0.0642	0.000 659	0.241 251	0.241 910	0.000	167.368	171.784	0.0187	0.7269	0.7269
-35	0.0807	0.000 666	0.194 732	0.195 398	4.416	165.222	174.076	0.0371	0.7027	0.7214
-30	0.1004	0.000 672	0.158 703	0.159 375	8.854	163.037	176.352	0.0552	0.6795	0.7165
-25	0.1237	0.000 679	0.130 487	0.131 166	13.315	160.810	178.610	0.0730	0.6570	0.7121
-20	0.1509	0.000 685	0.108 162	0.108 847	22.312	158.534	180.846	0.0906	0.6352	0.7082
-15	0.1826	0.000 693	0.090 326	0.091 018	26.851	156.207	189.058	0.1079	0.5936	0.7014
-10	0.2191	0.000 700	0.075 946	0.076 646	31.420	153.823	185.243	0.1250	0.5736	0.6986
-5	0.2610	0.000 708	0.064 255	0.064 963	36.022	151.376	187.397	0.1418	0.5542	0.6960
0	0.3086	0.000 716	0.054 673	0.055 389	40.659	148.859	189.518	0.1585	0.5351	0.6937
5	0.3626	0.000 724	0.046 761	0.047 485	45.337	146.265	191.602	0.1750	0.5165	0.6916
10	0.4233	0.000 733	0.040 180	0.040 914	50.058	143.586	193.644	0.1914	0.4983	0.6897
15	0.4914	0.000 743	0.034 671	0.035 413	54.828	140.812	195.641	0.2076	0.4803	0.6879
20	0.5673	0.000 752	0.030 028	0.030 780	59.653	137.933	197.586	0.2237	0.4626	0.6863
25	0.6516	0.000 763	0.026 091	0.026 854	64.539	134.936	199.475	0.2397	0.4451	0.6848
30	0.7449	0.000 774	0.019 855	0.020 641	69.494	131.805	201.299	0.2557	0.4277	0.6834
35	0.8477	0.000 786	0.017 373	0.018 171	74.527	128.525	203.051	0.2716	0.4104	0.6820
40	0.9607	0.000 798	0.015 220	0.016 032	84.868	125.074	204.722	0.2875	0.3931	0.6806
45	1.0843	0.000 811	0.013 344	0.014 170	90.201	121.430	206.298	0.3034	0.3758	0.6792
50	1.2193	0.000 826	0.011 701	0.012 542	95.665	117.565	207.766	0.3194	0.3582	0.6777
55	1.3663	0.000 841	0.010 253	0.011 111	101.279	109.024	210.303	0.3355	0.3405	0.6760
60	1.5259	0.000 858	0.008 971	0.009 847	107.067	104.255	211.321	0.3518	0.3224	0.6721
65	1.6988	0.000 877	0.007 828	0.008 725	113.058	99.068	212.126	0.3683	0.3038	0.6697
70	1.8858	0.000 897	0.006 802	0.007 723	119.291	93.373	212.665	0.3851	0.2845	0.6667
75	2.0874	0.000 920	0.005 875	0.006 821	125.818	87.047	212.865	0.4023	0.2644	0.6631
80	2.3046	0.000 946	0.005 029	0.006 005	132.708	79.907	212.614	0.4201	0.2430	0.6585
85	2.5380	0.000 976	0.004 246	0.005 258	140.068	71.658	211.726	0.4385	0.2200	0.6526
90	2.7885	0.001 012	0.003 508	0.004 563	148.076	61.768	209.843	0.4579	0.1946	0.6444
95	3.0569	0.001 056	0.002 790	0.003 903	157.085	49.014	206.099	0.4788	0.1655	0.6319
100	3.3440	0.001 113	0.002 045	0.003 242	168.059	28.425	196.484	0.5023	0.0742	0.6064
105	3.6509	0.001 197	0.001 364	0.001 098	0.002 462	174.920	0.151	175.071	0.5322	0.0004
110	3.9784	0.001 792	0.000 005	0.001 797					0.5651	0.5655
112	4.1155	0.001 792								

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Superheated Freon-12

Temp. °C	$\frac{v}{m^3/kg}$	$\frac{k}{kJ/kg}$	$\frac{s}{kJ/kg\ K}$	$\frac{v}{m^3/kg}$	$\frac{k}{kJ/kg}$	$\frac{s}{kJ/kg\ K}$	$\frac{v}{m^3/kg}$	$\frac{k}{kJ/kg}$	$\frac{s}{kJ/kg\ K}$			
	0.05 MPa	0.10 MPa	0.15 MPa	0.20 MPa	0.25 MPa	0.30 MPa	0.40 MPa	0.50 MPa	0.60 MPa	0.70 MPa	0.80 MPa	0.90 MPa
-20.0	0.41 857	181.042	0.7912	0.167 701	179.861	0.7401	0.114 716	184.619	0.7318			
-10.0	0.556 227	186.757	0.8193	0.175 222	185.707	0.7628	0.119 866	190.560	0.7543			
0.0	0.578 508	192.567	0.8350	0.182 647	191.528	0.7849	0.124 932	196.762	0.7763			
10.0	0.584 716	198.471	0.8562	0.189 994	197.628	0.8064	0.129 939	202.927	0.7977			
20.0	0.598 863	204.469	0.8770	0.197 277	203.707	0.8275	0.134 873	209.160	0.8186			
30.0	0.612 959	210.557	0.8974	0.204 506	209.866	0.8482	0.139 768	215.463	0.8390			
40.0	0.627 012	216.733	0.9175	0.211 691	216.104	0.8684	0.144 625	221.835	0.8591			
50.0	0.641 030	222.997	0.9372	0.218 899	222.421	0.8883	0.149 450	228.277	0.8787			
60.0	0.655 017	229.544	0.9565	0.225 955	228.815	0.9078	0.154 247	234.789	0.8980			
70.0	0.668 978	235.774	0.9755	0.233 044	235.285	0.9269	0.159 020	241.371	0.9169			
80.0	0.682 917	242.282	0.9942	0.240 111	241.829	0.9457	0.163 774	248.020	0.9354			
90.0	0.696 858	248.868	1.0126	0.247 159	248.446	0.9642						
0.0	0.088 608	189.669	0.7320	0.069 752	188.644	0.7139	0.057 150	187.583	0.6984			
10.0	0.092 550	195.878	0.7545	0.073 024	194.969	0.7366	0.059 984	194.034	0.7216			
20.0	0.096 418	202.135	0.7760	0.076 218	201.322	0.7587	0.062 734	200.490	0.7440			
30.0	0.100 228	208.446	0.7972	0.079 350	207.715	0.7801	0.065 418	206.969	0.7658			
40.0	0.103 989	214.814	0.8178	0.082 431	214.153	0.8010	0.068 049	213.480	0.7869			
50.0	0.107 710	221.245	0.8381	0.085 470	220.642	0.8214	0.070 635	220.030	0.8075			
60.0	0.111 597	227.735	0.8578	0.088 474	227.185	0.8413	0.073 185	226.627	0.8276			
70.0	0.115 055	234.291	0.8772	0.091 449	233.785	0.8608	0.075 705	233.273	0.8475			
80.0	0.118 690	240.910	0.8962	0.094 398	240.443	0.8800	0.078 200	239.971	0.8665			
90.0	0.122 304	247.595	0.9149	0.097 327	247.160	0.8987	0.080 675	246.723	0.8855			
100.0	0.125 901	254.339	0.9332	0.100 238	253.936	0.9171	0.083 127	253.530	0.9038			
110.0	0.129 483	261.147	0.9512	0.103 134	260.770	0.9352	0.085 566	260.391	0.9220			
20.0	0.045 836	198.762	0.7199	0.035 646	196.935	0.6999	0.030 422	202.116	0.7063			
30.0	0.047 971	205.428	0.7423	0.037 464	203.814	0.7250	0.031 966	209.154	0.7291			
40.0	0.050 046	212.095	0.7639	0.039 214	210.656	0.7452	0.033 450	216.141	0.7511			
50.0	0.052 072	218.779	0.7849	0.040 911	217.484	0.7667	0.034 887	225.104	0.7723			
60.0	0.054 059	225.488	0.8054	0.042 565	224.315	0.7875	0.036 285	230.062	0.7929			
70.0	0.056 014	232.230	0.8253	0.044 184	231.161	0.8077	0.037 655	237.027	0.8129			
80.0	0.057 941	239.012	0.8448	0.045 774	238.051	0.8275	0.038 995	244.009	0.8324			
90.0	0.059 846	245.837	0.8638	0.047 340	244.932	0.8467	0.040 516	251.016	0.8514			
100.0	0.061 751	252.707	0.8825	0.048 886	251.869	0.8656	0.041 619	258.053	0.8700			
110.0	0.063 600	259.624	0.9008	0.050 415	258.845	0.8840	0.042 907	265.124	0.8882			
120.0	0.065 455	266.590	0.9187	0.051 929	265.862	0.9021	0.044 181	272.231	0.9061			
130.0	0.067 298	273.605	0.9364	0.053 430	272.925	0.9198						
40.0	0.036 761	207.580	0.7148	0.022 830	205.924	0.7016	0.019 744	204.170	0.6982			
50.0	0.028 100	214.745	0.7373	0.024 068	213.290	0.7248	0.020 912	211.765	0.7131			
60.0	0.029 387	221.054	0.7590	0.025 247	220.558	0.7469	0.022 012	219.212	0.7358			
70.0	0.030 632	228.931	0.7799	0.026 380	227.766	0.7682	0.023 062	226.564	0.7575			
80.0	0.031 843	235.997	0.8002	0.027 477	234.941	0.7888	0.024 072	233.856	0.7765			
90.0	0.033 027	243.066	0.8199	0.028 545	242.101	0.8088	0.025 051	241.113	0.7987			
100.0	0.034 189	250.146	0.8392	0.029 588	249.260	0.8283	0.026 005	248.355	0.8184			
110.0	0.035 352	257.247	0.8579	0.030 612	256.428	0.8472	0.026 937	255.593	0.8376			
120.0	0.036 458	264.374	0.8763	0.031 619	263.613	0.8657	0.027 851	262.839	0.8562			
130.0	0.037 572	271.531	0.8943	0.032 612	270.820	0.8838	0.028 751	270.100	0.8745			
140.0	0.038 673	278.720	0.9119	0.033 592	278.055	0.9016	0.029 639	277.381	0.8923			
150.0	0.039 764	285.946	0.9292	0.034 563	285.320	0.9189	0.030 515	284.687	0.9092			
50.0	0.018 566	210.162	0.7021	0.014 483	206.661	0.6812	0.012 579	211.457	0.6876			
60.0	0.019 410	217.810	0.7254	0.015 463	214.805	0.7060	0.013 448	219.822	0.7123			
70.0	0.020 997	225.319	0.7476	0.016 368	222.687	0.7293	0.014 247	227.891	0.7355			
80.0	0.021 341	232.739	0.7689	0.017 221	230.398	0.7514	0.014 997	235.766	0.7575			
90.0	0.022 251	240.101	0.7895	0.018 052	237.995	0.7727	0.015 710	243.512	0.7785			
100.0	0.023 153	247.450	0.8094	0.018 812	245.818	0.7931	0.016 393	251.170	0.7988			
110.0	0.023 993	254.743	0.8287	0.019 567	252.935	0.8129	0.017 053	258.770	0.8183			
120.0	0.024 835	262.053	0.8475	0.020 301	260.441	0.8320	0.017 695	266.334	0.8373			
130.0	0.025 661	269.369	0.8659	0.021 018	267.875	0.8507	0.018 321	273.877	0.8558			
140.0	0.026 474	276.699	0.8839	0.021 721	275.307	0.8689	0.018 934	281.411	0.8758			
150.0	0.027 275	284.047	0.9015	0.022 412	282.745	0.8867	0.019 535	288.946	0.8914			
				0.023 093	290.195	0.9041	0.019 535	287.201	0.8820			
70.0	0.011 203	216.650	0.6959	0.009 406	215.049	0.6704	0.008 704	218.859	0.6909			
80.0	0.011 984	225.177	0.7204	0.010 187	222.198	0.7057	0.009 406	228.056	0.7166			
90.0	0.012 698	233.890	0.7433	0.010 884	230.835	0.7298	0.010 035	236.760	0.7402			
100.0	0.013 266	241.397	0.7651	0.011 526	239.155	0.7524	0.010 615	245.154	0.7624			
110.0	0.014 000	249.264	0.7859	0.012 126	247.264	0.7759	0.011 159	253.341	0.7835			
120.0	0.014 608	257.035	0.8059	0.012 697	255.228	0.7944	0.011 676	261.884	0.8037			
130.0	0.015 195	264.742	0.8263	0.013 244	263.094	0.8141	0.012 172	269.327	0.8252			
140.0	0.015 765	272.406	0.8440	0.013 772	270.891	0.8332	0.012 651	277.201	0.8420			
150.0	0.016 320	280.044	0.8623	0.014 284	278.642	0.8518						