

National Exams May 2019

17-Pet-B2, Natural Gas Engineering

3 hours duration

NOTES:

1. If doubt exists as to the interpretation of any question, the candidate is urged to submit with the answer paper, a clear statement of any assumptions made.
2. This is an OPEN BOOK exam.
3. Any non-communicating calculator is permitted.
4. Seven (7) questions constitute a complete exam paper.
5. The first five questions as they appear in the answer book will be marked.
6. Each question is of equal value.
7. Pay close attention to units, some questions involve oilfield units, and these should be answered in the field units. Questions that are set in other units should be answered in the corresponding units.

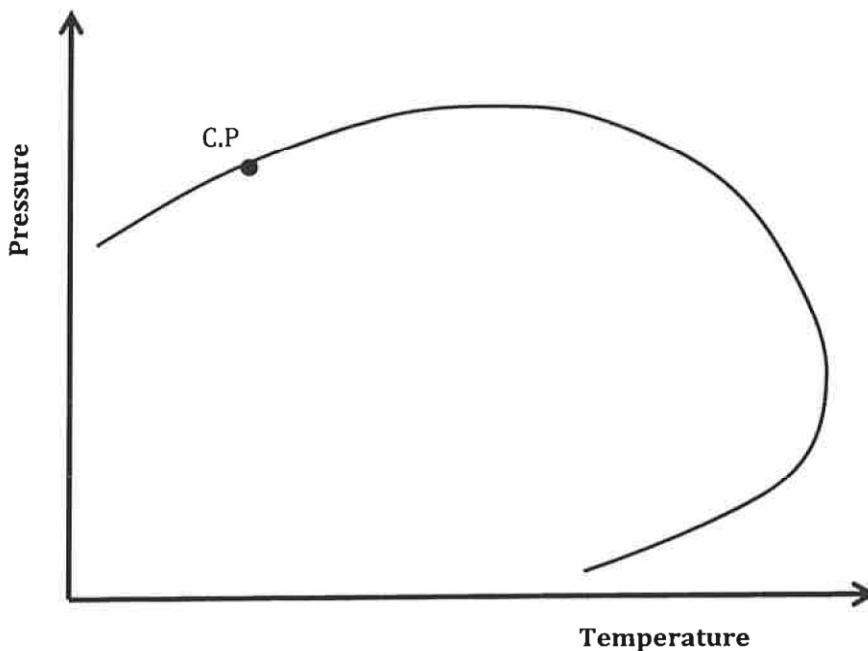
Question Number I (10 Marks)

- a) Define briefly and concisely the following terms that are commonly used in natural gas recovery and processing:
- i. Gas reserve
 - ii. Effective permeability
 - iii. Relative permeability
 - iv. Interfacial tension
 - v. Resistivity index
 - vi. Capillary pressure
 - vii. Gas volume formation factor
 - viii. Wettability

Question Number II (10 Marks)

The following graph presents the phase envelope of a typical natural gas reservoir. Redraw the graph on your exam book and show the followings on the graph:

- a. Liquid phase region
- b. Gas phase region
- c. Two phase region (liquid and vapour)
- d. Bubble point line
- e. Dew point line
- f. Critical point
- g. Critical temperature point



Question Number III (10 Marks)

A gas mixture with compositions listed in the table below is held at a pressure of 10.19 MPa and a temperature of 271 K. Calculate:

- The molar volume (m^3/kmol) and the density (kg/m^3) of the mixture using the Kay's rule and the Katz correlation?
- The molar volume (m^3/kmol) and the density (kg/m^3) of the mixture using the approach of Wichert and Aziz to adjust for CO_2 and H_2S ?
- If the gas flows to a processing facility at a rate of $2800 \times 10^3 \text{ Sm}^3/\text{d}$, what is the mass flow of the gas in kg/s?

Component	Mole fraction (y)
N_2	0.0050
CO_2	0.2250
H_2S	0.1000
C_1	0.6610
C_2	0.0070
C_3	0.0015
i- C_4	0.0003
n- C_4	0.0002
	1.0000

Question Number IV (10 Marks)

- Using the results from Part b in Question III above, and the flow rate from Part c, what is the actual volumetric flow rate of the gas at the plant in m^3/s ?
- Determine the standard gas flow rate in MMSCFD if the standard conditions are 60 °F and 14.69 psia?

Question Number V (10 Marks)

Use the following information in the table below to calculate the gas flowrate in a horizontal pipeline. You are free to use any flow equation (like Basic, Weymouth, Panhandle A, Panhandle A and AGA equations).

Variable	Value	Units
P_1 = inlet absolute pressure	7000	kPa
P_2 = outlet absolute pressure	5600	kPa
PSC = standard pressure	101.325	kPa
TSC = standard temperature	15	°C
d = inside pipe diameter	0.3049	m
y = gas relative density	0.7	
L = pipe length	50,000	m
T_m = mean temperature of the line	27	°C
z_m = mean compressibility factor	0.79	
μ = gas viscosity	0.0114	cP
E = pipeline efficiency	0.92	
ϵ = roughness	40	μm

Question Number VI (10 Marks)

A hydrocarbon gas-liquid mixture is flowing horizontally in a pipe having an inside diameter of 200 mm and length of 10 km. The liquid density is 840 kg/m^3 and the gas density is 35 kg/m^3 . The liquid volumetric flow rate is $50 \text{ m}^3/\text{h}$ and the gas flow rate is $500 \text{ m}^3/\text{h}$ measured at standard conditions. The temperature and pressure at the inlet of the pipeline are 40°C and 5000 kPa, respectively. The gas average compressibility and molecular weight is 0.85 and 22 kg/kmol, respectively. Estimate the expected flow regime using the method developed by Gregory-Aziz-Mandhane?

Question Number VII (10 Marks)

You have obtained the following dray-gas reservoir data estimated from subsurface maps, core analysis, well tests and fluid samples obtained at several wells. Using the volumetric method, estimate the original gas in place?

p_i	2000 psia
T	160°F
A	8000 acres
ϕ	20%
S_{wi}	22%
h	8 ft
z_i	0.88

Formula Sheet

Gas properties:

$$M_a = \sum y_i M_i, \quad \text{where } y \text{ is mole fraction and } M \text{ is molecular weight in lb_{mass}/lb_{mole},}$$

$$\gamma_g = \frac{M_a}{M_{air}}, \quad \gamma_g \text{ is gas specific gravity (Air=1),}$$

$$T_{pc} = 169.2 + 349.5\gamma_g - 74.0\gamma_g^2, \quad T_{pc} \text{ is the pseudo critical temperature in degree R,}$$

$$p_{pc} = 756.8 - 131.0\gamma_g - 3.6\gamma_g^2, \quad p_{pc} \text{ is the pseudo critical pressure in psia,}$$

$$\text{Correction for N}_2, \text{H}_2\text{S, and CO}_2: T'_{pc} = T_{pc} - 80y_{CO_2} + 130y_{H_2S} - 250y_{N_2}$$

$$\text{Correction for N}_2, \text{H}_2\text{S, and CO}_2: p'_{pc} = p_{pc} + 440y_{CO_2} + 600y_{H_2S} - 170y_{N_2}$$

$$T_r = \frac{T}{T'_{pc}}, \quad p_r = \frac{p}{p'_{pc}}$$

T_r and p_r are reduced pseudo critical temperature and pressure, respectively.

$$\rho = \frac{pM}{ZRT} \quad \text{where } \rho \text{ is gas density in lb_{mass}/ft}^3, p \text{ in psia, } T \text{ in R, } M \text{ is in lb_{mass}/lb_{mole}, } R=10.732 \text{ psi-ft}^3/(\text{lb mol} \cdot ^\circ\text{R})$$

$$\text{Gas formation volume factor, } B_g = 0.02827 \frac{ZT}{p} \quad \text{in } \frac{\text{ft}^3}{\text{SCF}}, \quad \text{where } p \text{ in psia, } T \text{ in } ^\circ\text{R.}$$

Standard condition: $T_{sc}=60 \text{ } ^\circ\text{F}$, $p_{sc}=14.7 \text{ psia}$.

Pipeline flow capacity equations:

$$q_{sc} = 5.634 \left(\frac{T_{sc}}{P_{sc}} \right) \sqrt{\frac{(p_1^2 - p_2^2)d^5}{f\gamma_g Z_{av} TL}} \quad \text{where } T \text{ in } ^\circ\text{R, } d \text{ in inch, } L \text{ in ft, } q \text{ in MSCFD.}$$

$$N_{Re} = 710.39 \frac{P_{sc}}{T_{sc}} \frac{\gamma_g q_{sc}}{\mu d} \quad q \text{ in MSCFD, viscosity in cP, } d \text{ in inches.}$$

Decline curve analysis

$$\text{Exponential decline: } q = q_i e^{-Dt},$$

$$\text{Harmonic decline: } q = q_i / (1 + Dt)$$

$$\text{Hyperbolic decline } q = q_i (1 + bDt)^{-1/b}$$

$$\text{Cumulative production } G_p = \int q dt$$

where q is rate in SCFD, G_p is the cumulative production in SCF, t is time in day, D is the decline rate in 1/day and subscript i stands for the initial condition.

Transient flow equations in field units:

$$\psi(r, t) = \psi_i - \frac{1.422 q_{sc} T}{kh} p_D, \quad \eta = \frac{6.33k}{\phi \mu_i c_i}, \quad t_D = \frac{\eta t}{r_w^2}$$

$$p_D = \frac{1}{2} (\ln t_D + 0.809) \quad \text{only if } t_D > 100,$$

$$\psi(r,t) = \psi_i - \frac{1.422 q_{sc} T}{kh} p_D$$

where q_{sc} is gas rate in MSCFD, ψ is the real gas pseudo pressure in psi^2/cp , ϕ is porosity, t is time in day, t_D is the dimensionless time, k is permeability in Darcy, h is formation thickness in ft, r is radius in ft, p is pressure in psia, c is the isothermal compressibility in psi^{-1} , μ is the gas viscosity in cP, T is temperature in R, S is skin factor, and p_D is the dimensionless pressure. The subscript i denotes the initial condition.

Gas wells drawdown test

Slope of the semilog-plot: $m = \frac{1637 q_g T}{kh}$, q_g is in MSCFD, T is °R, k in mD, h in ft.

Test skin factor: $S' = 1.151 \left(\frac{\psi_i - \psi(\Delta t = 1\text{hr})}{|m|} - \log \left(\frac{k}{\phi \mu_i c_n r_w^2} \right) + 3.23 \right)$, where S' is the test skin factor, c is the gas isothermal compressibility in psi^{-1} , μ is the gas viscosity in cP, and ϕ is porosity

True skin factor: $S' = S + Dq$, where D is the non-Darcy or turbulent factor in 1/MSCFD

Gas wells build up test

Slope of the semilog-plot: $m = \frac{1637 q_g T}{kh}$, q_g is in MSCFD, T is °R, k in mD, h in ft.

Test skin factor: $S' = 1.151 \left(\left(\frac{\psi_{1\text{hr}} - \psi_{wf}(\Delta t = 0)}{m} \right)_{1\text{hr}} - \log \left(\frac{k}{\phi \mu c_r r_w^2} \right) + 3.23 \right)$ where S' is the test

skin factor, c is the gas isothermal compressibility in psi^{-1} , μ is the gas viscosity in cP, and ϕ is porosity

Gas wells deliverability equation:

$q = C(\bar{p}^2 - p_{wf}^2)^n$ where \bar{p} is the average reservoir pressure, and p_{wf} is the stabilized flowing wellbore pressure, q is the gas production rate, C is the coefficient of the equation in any consistent systems of unit and n is an exponent.

Conversion Factors

1 m³ = 6.28981 bbl = 35.3147 ft³

1 acre = 43560 ft²

1 ac-ft = 7758 bbl

1 Darcy = 9.869233×10^{-13} m²

1 atm = 14.6959488 psi = 101.32500 kPa = 1.01325 bar

1 cP = 0.001 Pa·sec

1 m = 3.28084 ft = 39.3701 inch = 0.000621371 mile

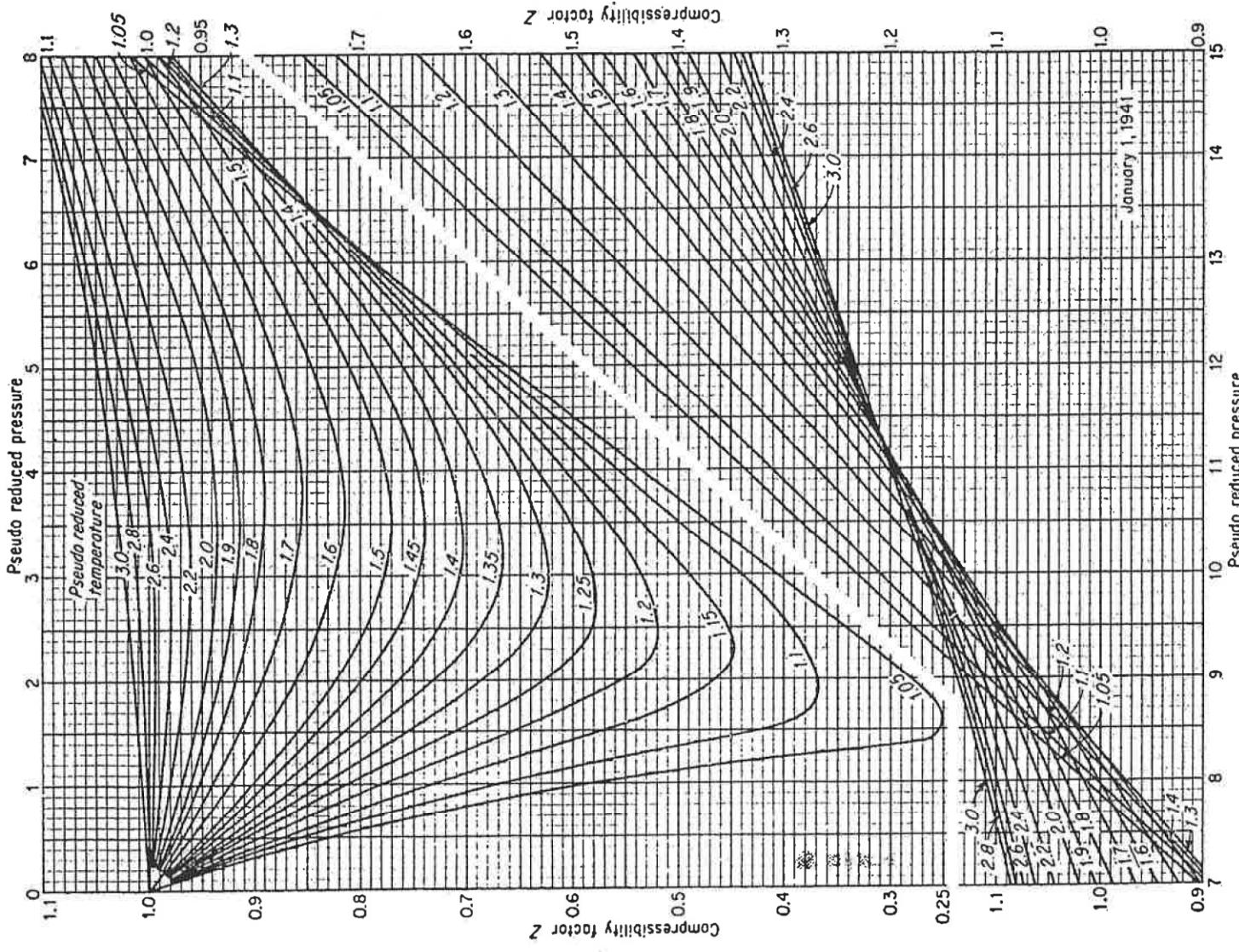
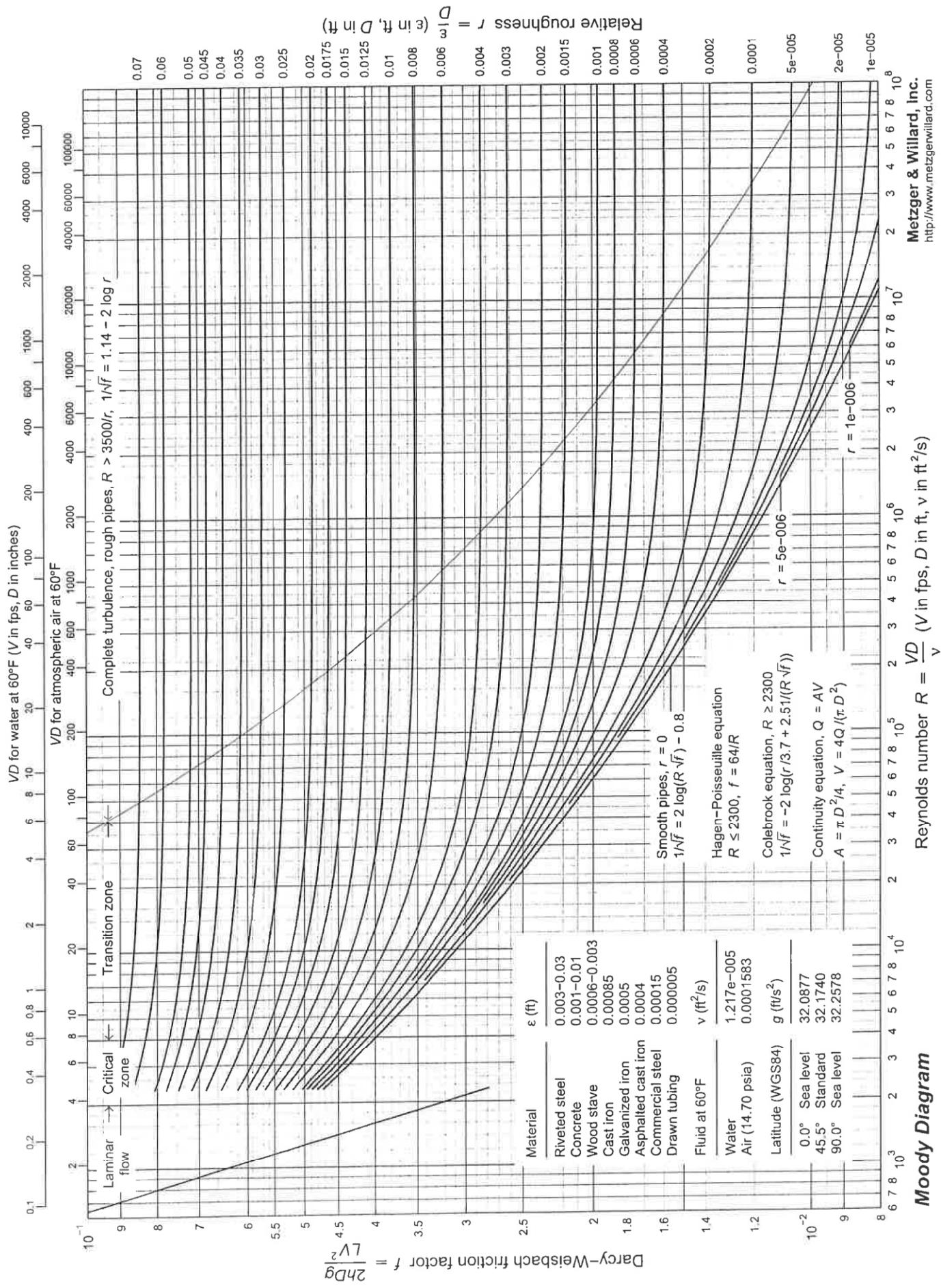


Fig. 4-16. Compressibility factor for natural gases. (Standing and Katz, 4-87. Courtesy AIME.)



Moody Diagram